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#### A Study on the Soil Pollution in Lead Industry

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This study is to find out which the elements could cause the pollution in the soil, and to what the degree of the pollution has been progressed so far, with samples of the soils from such work places as 6 lead storage battery factories, a primary lead smelting factory, and a litharge factory, and finally from the area that is considered not to have been polluted by any industrial activities. The study variables are the number of employees, the date of foundation of workplace, the amount of use of lead, the size of the land, the size of facilities, and the capacity of dust collector. We inspected the accuracy and the precision of the portable X-ray Fluorescence which has been rarely used in this nation. Followings are the details of our inspection.

- 1. For the verification of the accuracy and the precision, we prepared three different concentrations. For the accuracy, we had 219.6 % collection rate from low concentration, 97.8 % from middle concentration, and 101.4% form high concentration. We found the variation modules by concentration, for the precision, to be 23.1 % from low concentration, 1.91 % from middle concentration, and 0.66 % from high concentration, the result of which shows us that the portable X-ray fluorescence's accuracy and precision are somewhat low with low concentration, and high with middle or high concentration.
- 2. The lowest level of pollution in workplace caused by lead was that of D company's with its average concentration was 182±2.512mg/kg the highest, C company's with 72,069 ±2.548mg/kg.
  - 3. We studied the association the date of foundation of

workplace, the amount of use of lead, the size of the land, the rate of the building occupation of the land, lead amount in the soil, the capacity of the dust collector and devided the capacity of the dust collector by the size of the building to find out the relation between the dust collector's affection per unit area and the rate of lead in the soil.

From this study, which indicate the older the date of foundation of workplace, the smaller using lead, occupation of the land, the building occupation of the land, the capacity of the dust collector, the rate of the dust collector's affection per unit area.

4. We made a comparitive study of only the groups of the storage battery company in the same way as above. From this study, we had almost the same result as we did from the study on the whole lead-related workplaces; the only different results we got from the variation of the amount the number of employees, which indicates the bigger the number of employees, the less soil pollution. it shows and the result is statistically worthy of attention(p<0.05).

Is judged that use about utilization because it analytical accuracy and precision are high and have a lot of advantages about free medical care pretreatment and sample ore that analysis is difficult with galena specially in concentration more than constant level to inflect portable XRF in measurement about lead concentration among soil in place of business and estimation.

Key Words: Portable X-ray Fluorescence, Pollution in the soil

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2 I. 서 론 XRF ( 1992), 2 (Litharge; lead oxide) 가 , 1999). ( Ⅱ. 연구대상 및 방법 1. 연구대상 및 시기 6 1 가 5-20 2003 3 12 75 , 6가 가 ) ( ( ), 28 ), (PCB), 1 가 2 , 가 XRF(portable X-ray fluorescence) 2. 시료채취 및 분석방법 2001). 가 1) 시료채취 및 준비 가 EPA Method 6200 (AAS) 가 (Zenz, (ICP) 1994). EPA Method 6200 가, X-XRF 10cm 가 가 가 150 가 60 mesh가 1 가 (Pirkle (2001)1994), XRF sample cup 1/4 mil mylar film 가 1960 ( , 1968). 1972

가 .

1983

1986

8

( , collar)

mylar film

1/4 mil

3

Table 1. 나업장의 일반적인 현황

	( )		( )	(ton/ )	(m²)	$(m^2)$	(m³/min)
A	630		1975	3,250	92,423	43,836	20,000
В	110		1995	3,119	33,058	11,570	5,841
C	64		1977	500	18,559	13,957	-
D	81		1997	913	62,810	18,512	10,720
E	360		1985	3,760	51,570	14,545	26,660
F	300		1988	2,000	46,358	28,384	13,680
G	87		1993	450	18,003	12,872	3,512
Н	720	1	1988	19,000	648,939	243,425	98,299

Multiple Com-Duncan parison Test 2) (% 가 recovery rate) Bias <sup>109</sup>Cd X-ray Fluorescence (XL-309 lead analyser, (NIOSH, 1995; Taylor, 1987). NITON Corp, Bedford, Massachusetts) 회수율(%) = <u>분석량</u> × 100 . XRF **EPA** 1) METHOD 6200 3 XRF = |[-<sup>(분<u>석량</u> - 첨<u>가량</u>)]×100| 첨가량</sup> Test Platform( 4) sample cup 2) 3. 자료분석 (coefficient of variation, CV) (overall SAS version 8.0(Statical CV, CV<sub>pooled</sub>) . CV Analysis System)  $CV_{\text{pooled}} \\$ (NIOSH, 1995; OSHA, 1990). ANOVA(Analysis of Variance) test 3)

Table 2. 표준시료의 분석결과

	(18.9mg/kg)	(1,162.0mg/kg)	(5,532.0mg/kg)
1	26.4	1,100.0	5,560.0
2	29.0	1,110.0	5,560.0
3	34.3	1,120.0	5,580.0
4	38.0	1,130.0	5,580.0
5	41.0	1,140.0	5,600.0
6	45.1	1,140.0	5,620.0
7	46.0	1,141.0	5,630.0
8	46.4	1,150.0	5,640.0
9	53.9	1,160.0	5,650.0
10	54.4	1,170.0	5,660.0

$$CVpooled = \left[ -\frac{\sum_{i=1}^{n} f_{i}(CV_{i})^{2}}{\sum_{i=1}^{n} f_{i}} \right]^{-\frac{1}{2}}$$

$$I = n . 7$$

$$CV_{i} = i$$

$$f_{i} = ( -1)$$

# Ⅲ. 연구결과

## 1. 휴대용 XRF의 정확도 및 정밀도

**XRF** 

가

(1,162.0mg/kg) 97.8%, (5,532.0mg/kg) 101.4% 가 (CV %) (18.9 mg/kg)23.1%, (1,162 mg/kg)1.91%, (5,532.0 mg/kg) 0.66% ( ) **XRF** (CV %)

**START** 1 Follow manufacturers manual for operation of FPXRF instrumentation 2 Type of analysis in situ intrusive mode 3 Remove debris from 4 Collect sample from soil surface and level a 4×4 inch square of surface. if necessary. Tap soil soil to increase density and compactness. 3 Perform analysis No Follow preparation Sample procedure achieve homogenization before drying? your DQOs. 4 Thoroughly mix sample in a beaker or plastic bag. Monitor homogenization with sodium fluorescein dye. 5 Dry 20-50 grams of sample for 2-4 hours at a temp. no greater than 150 6 Ground sample until 90% of original sample passes through a 60-mesh sieve 6 Place sample in polyethylene sample cup and perform analysis

#### METHOD 6200

STOP

Field portable X-ray Fluorescence spectrometry for the determination of elemental concentrations in soil and sediment

Fig 1. EPA method 6200



Fig 2. Sample cup with 1/4 mil mylar film



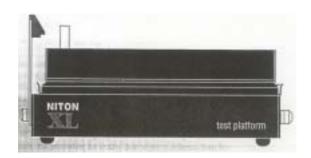
Fig 3. Sample cup with cap



The NITON test Platform



Placing the NITON on the NITON Test Platform



Firmly latching the NITON flat against the NITON Test Platform

Fig 4. NITON Test Platform 시료분석 과정

( 3).

# 2. 토양 중 납 농도에 관한 결과

4

mg/kg mg/kg	100mg/l , A 218±2.316mg/kg	20 G 230±3.597 100
	가가	5 D

. 가가		D
182±2.512mg/kg		, C 가
72,069±2.548mg/kg	가	
. 8	1	
		400
mg/kg	,	
		1,000mg/kg
6		

가 100	, 101	300
301	3	
	•	

가				
7				
	500ton	,	501	ton
2,000ton	, 2,001to	on		10,000ton
10	,001ton		4	
				(p<0.05).
8				

	30,0	00 <b>m</b> ²	, 30,001 m <sup>2</sup>
	50,000 m <sup>2</sup>	50,	001 <b>m</b> ²
3			

6 . . . . . .

Table 3. XRF 기기의 정확도 및 정밀도

(M)	(SD)	Bias  (%)	(Rec %)	(CV %)	(CVpooled %)
41.5	9.58	119.7	219.6	23.08	
1,136.1	21.73	2.2	97.8	1.91	13.3
5,608.0	37.06	1.4	101.4	0.66	

Table 4. 나업장이 소재한 지역의 토양 중 납 농도

				(
		±		
A	4	$218 \pm 2.316$	59	495
В	4	$83 \pm 1.073$	67	119
C	4	$74 \pm 1.197$	49	106
D	4	$69 \pm 1.105$	58	113
E	4	$77 \pm 1.419$	49	118
F	4	$57 \pm 1.323$	43	88
G	4	$230 \pm 3.597$	73	1380

Table 5. 시업장별 토양 중 납 농도의 평균

				( : mg/kg)
		<u>+</u>		
A	20	$3,218 \pm 2.649$	1,080	43,233
В	8	$1,598 \pm 10.646$	86	251,333
C	5	$72,069 \pm 2.548$	23,400	186,000
D	8	$182 \pm 2.512$	67	913
E	7	$667 \pm 2.622$	281	3,296
F	8	$18,932 \pm 8.635$	515	239,000
G	8	$9,793 \pm 8.660$	2,103	665,333
H	11	$5,685 \pm 7.054$	364	397,000

Table 6. 근로자수와 토양 중 납 농도외의 관계

( )		± (mg/kg)
100	21	$3,454 \pm 17.845$
101 - 300	16	$5,500 \pm 12.574$
301	38	$2,839 \pm 4.434$
p value		0.599

Table 7. 시업장의 납 사용량과 토양 중 납 농도와의 관계

(ton/ )		$\pm$ (mg/k	g) Duncan grouping(r)
500	13	$21,102 \pm 7.448$	a
501 - 2,000	16	$1,858 \pm 17.867$	b
2,001 - 10,000	35	$2,002 \pm 4.465$	b
10,001	11	$5,685 \pm 7.055$	ab
F value			5.10
p value			0.003
0.05		. 가	

Table 8. 나업장의 전체 토지면적과 토양 중 납 농도와의 관계

(m²)		± (mg/kg)	Duncan grouping
30,000 30,001 - 50,000	13 16	$21,102 \pm 7.448$ $5,500 \pm 12.574$	a b
50,001	46	$1,761 \pm 5.574$	b
F value p value			8.53 0.0005

Table 9. 나업장 건물면적과 토양 중 납 농도와의 관계

$(m^2)$		± (mg/kg)
15,000	28	$4,257 \pm 11.247$
15,001 - 50,000	36	$2,521 \pm 7.676$
50,000	11	$5,685 \pm 7.7.055$
p value		0.457

가 (p<0.05), 10 30,000m<sup>2</sup> 50,001 m<sup>2</sup> 가 9 2.00 , 2.01 3.00 3.01 11 3 15,000 m<sup>2</sup> , 15,001 m<sup>2</sup> 5,000 m<sup>3</sup>/min , 5,001 m<sup>3</sup> 50,000m<sup>2</sup> 50,001 m<sup>2</sup> 10,000 m<sup>3</sup>/min , 10,001 m<sup>3</sup> /min 20,000 m<sup>3</sup>/min /min 20,001 (p<0.05). m³/min

Table 10. 나업장 건물면적당 토지면적과 토양 중 납 농도와의 관계

/ ( )		± (mg/kg)	Duncan grouping
2.00	21	$20,248 \pm 7.479$	a
2.01 - 3.00	39	$3,273 \pm 5.181$	b
3.01	15	$334 \pm 3.088$	c
F value			26.36
p value			0.0001

Table 11. 사업장의 집진기용량과 토양 중 납 농도와의 관계

(m <sup>3</sup> /min)		± (mg/k	g) Duncan grouping
5,000	13	$21,102 \pm 7.448$	a
5,001 - 10,000	8	$1,598 \pm 10.647$	b
10,001 - 20,000	36	$2,521 \pm 7.676$	b
20,001	18	$2,471 \pm 6.894$	b
F value			4.23
p value			0.0083

Table 12. 선물 면적당 집진기용량 비율과 토양 중 납 농도와의 관계

/ ( )		± (mg/kg)	Duncan grouping
0.30	13	$21,102 \pm 7.448$	a
0.31 - 0.49	39	$5,434 \pm 5.343$	b
0.50	23	$576 \pm 5.933$	c
F value			19.92
p value			0.0001

(p<0.05). . 14 13 가 100 12 300 , 301 , 101 가 1980 , 1981 1990 3 가 , 1991 1995 1996 4 (p<0.05), 0.30 , 0.31 0.50 3 가 0.49 (p<0.05). Ⅳ. 고 찰 (p<0.05).  $1 \text{m}^2$ 가 가

Table 13. 사업장 설립연도와 토양 중 납 농도와의 관계

( )		± (mg/kg)	Duncan grouping
1980	25	5,992 ± 4.873	a
1981 ~ 1990	26	$4,624 \pm 8.850$	a
1991 ~ 1995	16	$3,956 \pm 10.799$	a
1996	8	$182 \pm 2.5192$	b
F value			6.95
p value			0.0004

Table 14. 축전지 제조회사의 근로자 수와 토양 중 납 농도와의 관계

( )		± (mg/kg)	Duncan grouping
100	21	$18,933 \pm 1.058$	b
101 - 300	8	$3,454 \pm 8.635$	a
301	27	$2,140 \pm 3.267$	b
F value			3.31
p value			0.0443

```
40-313mg/kg
                                                           76mg/kg
           가
                                                                                                      가
                                                  11-54mg/kg
                                                       26mg/kg
                               가
                                                       (2000)
                                        가가
                                                                                                           가
                                                         가
                                                                                                         가
                                                                                                 가
                                                                                        가
                                              1997
                                                     4.741mg/kg, 1998
                        가
                                      8.906mg/kg
                                                     가 5.482mg/kg,
                                      7.146mg/kg,
                                                        5.177mg/kg,
    가
                                      7.461mg/kg
                                      57-230mg/kg
                                                        182-72,069mg/kg
  小林 (1973)
                         950mg/kg
                                                                  가 2.6~
                                                                                               가
                                                                                                          XRF
               100mg/kg
                                      973.9
                                                                                                  가
                         (1997)
                                                                     가
                                                                     가
        3.5km
                                                                                               가
         가 11.32-43.94mg/kg
                                                                                         Ⅴ. 결 론
                                                                                      1
                             가
TEL(Tetra-ethyl lead),
                                                                                                      가
                                                                                            X-ray fluorescence(XRF)
17mg/kg(Rose , 1979),
                    20ppm(
1982)
         Kloke (1979)가
        (thresholds) 100mg/kg
100mg/kg
                                                                             XRF
                 (1996)
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10 . . . . . . . .

1. XRF		
3		1999; 5(1): 49-65
·	가 .	. 2002
가		. 2002
(18.9mg/kg) 219.6%,		
(1,162.0mg/kg) 97.8%,	REFERENCES	2001
(5,532.0mg/kg) 101.4% ,		小林 純,森井ふじ,村本茂樹,中烏 進,涌
가 (CV %)	,	上佳子, 西崎日佐夫. 群馬縣安中市の
(18.9mg/kg) 23.1%,		亞鉛製錬所を中心とする土壤中の
(1,162.0 mg/kg) 1.91%,	. 1993; 30:	Cd, Pb, Zn の 分布にいて. 日本土肥
(5,532.0mg/kg) 0.66%	228-237	誌 1973; 44: 471-485
( ) (CV %)가	, .	Bernard BP, Becker CE. Environmental lead
		exposure and kidney. J Clin Tocicol
	2000; 6: 9-15	1988; 26: 1-34
2. 가	, , .	Clark S, Menrath W, Chen M, Roda S,
D 182±		Succop P. Use of a field portable X-ray
2.512mg/kg , 7\ C		fluorescence analyzer to determine the
72,069±2.548mg/kg .	1996; 17(2): 192-204	concentration of lead and other metals
3. , ,	, , , , , ,	in soil sample. Ann Agric Environ Med
,	, . X-ray	1999; 6: 27-32
( / ),	fluorescence high volume air sampler	EPA(Environmental Protection Agency).
,	2004	Analytical methods manual; Method
7L /	. 2004;	No. 6200, 1998
가 (	4(1): 71-76	Manahan SE. Fundamentals of enviro-
/ ) 가	, , , , , , , , , , , , , , , , , , ,	nmental chemistry. Lewis Publisher, 1993, p. 756
71	. A-Ray (ARI') 가.	Morley JC, Clark CS, Deddens JA, Ashely
,	2001; 11(3): 235-240	K, Roda S. Evaluation of portable
, ( / )		x-ray fluorescence instrument for the
, , ,	, , , .	determination of lead in workplace air
,		samples. Applied Occupational and
가	1994; 1(2):	Environmental Hygiene 1999; 14:
( / )	80-84	306-314
,		NIOSH. A NIOSH technical report;
(p<0.05).		Guidelines for air sampling and anal-
4.		ytical method developmental evalua-
		tion. Cincinnati, OH: DHHS (NIOSH)
	. 1992, p 9-12	Publication, 1995
	, , .	NITON Corporation. User's Guide. Version
		5.2. [Product Bulletin]. NITON
가	1997; 40(4):	Corporation, Bedford, MA. 1998
(p<0.05).	342-346	OSHA(Occupational Safety and Health
		Administration). Analytical methods
가 XRF	. , 2000	manual. Method No. 121, 1990
	, , , , , , ,	Pirkle JL, Brody DJ, Gunter Ew, Kramer
가		RA, Paschal DC, Flegal MM, Matte

TD. The decline in blood lead levels in the United States. JAMA 1994; 272(4): 284-291

Sterling DA, Lewis RD, Luke DA, Shadel BN. A portable X-ray fluorescence

instrument for analyzing dust wipe samples for lead; Evaluation with field samples. Environmental Research Section A 2000; 83: 174-179

Taylor JK. Quality assurance of chemical measurement. ed., Chelsea MI, Lewis

Publishers Inc., 1987

Zenz C. Occupational medicine. Year Book

Medical Publishers Inc., Chicago, 1994,
pp. 506-509